

Synthesis and Characterization of 1,3-Oxazepindion Derivatives via Diformylbenzene Imines as Precursors

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Abstract

In this paper, 1,3-oxazepindione derivatives [D] and [E] were prepared by reacting Schiff bases di Azomethine [A], [B], and [C] with maleic anhydride and succinic anhydride in the presence of dried benzene as a solvent. Schiff bases di Azomethine [A], [B], and [C] were prepared by reacting Terephthalaldehyde, Isophthalaldehyde, or o-Phthalaldehyde with 5 drops of glacial acetic acid as a catalyst and halo-aniline derivatives in absolute ethanol as a solvent. The prepared compounds were diagnosed and confirmed by using infrared and nuclear magnetic resonance spectroscopies. All the reactions of the prepared compounds were monitored by TLC.

Introduction:

Schiff bases were produced for the first time by the scientist Hugo Schiff in 1864 by thermal condensation between each of the ketones or aldehydes and the primary amine [1]. Schiff bases were distinguished by their biological activities against cancer [2], bacteria [3], and fungi [4]. Schiff bases had given other names, like Azomethine and Imine [1]. The term "heterocyclic compounds" refers to cyclic organic compounds that contain at least one different atom (or more) within the ring. And the most common different atoms found in the structure of heterocyclic rings are nitrogen, phosphorous, oxygen, and sulfur atoms [5]. The synthesis of heterocyclic rings is important due to their biological activities against bacteria, viruses, and inflammation. Some of these rings have been considered precursors for many important biological molecules [6]. For instance, vitamins, hemoglobin, chlorophyll, RNA, DNA, and numerous other compounds and Heterocyclic rings have further uses as insecticides and disinfectants [7]. Oxazepine derivatives are seven-member heterocyclic organic compounds that consist of a nitrogen atom, an oxygen atom, and five carbon atoms. Oxazepine derivatives are known for their high level of stability, due to the conjugation between the double bonds in the ring. There are three isomers of oxazepines: 1,4-oxazepine, 1,3-

oxazepine, and 1,2-oxazepine [8] [9]. Oxazepine derivatives are produced via a cyclo-addition reaction (2+5) between Schiff bases and cyclic anhydrides [10]. Oxazepine derivatives are active against bacteria [11], fungi [12], microbials [13], and tumors [14], according to previous studies and literature [15]. This paper seeks to synthesize Schiff Bases di Azomethine and use them as precursors for the preparation of new 1,3-oxazepindione derivatives.

EXPERIMENTAL PART

Materials:

Merck, Sigma-Aldrich, and Scharlau supplied all of the chemicals and solvents used in this study. Bruker Ascend™ 400 and Bruker Tensor 27 were used to record ¹³C-NMR, ¹H-NMR, and FT-IR spectra. The Stuart SMP-10 equipment was used to determine melting point temperatures.

General Method for Preparing Schiff Bases di Azomethine (A₁-A₃) (B₁) (B₂) (C₁) (C₂):

In a 100-mL reaction flask containing 0.001 moles (2g) of Terephthalaldehyde, Isophthalaldehyde, or *o*-Phthalaldehyde in 20mL of hot absolute ethanol with 5 drops of glacial acetic acid. 0.002 moles of dissolved aniline substitutions in 30 mL of hot absolute ethanol were added. For 3–4 hours, the reaction mixture was left to reflux. The precipitate was filtered once the reaction's completion was verified, and recrystallized using absolute ethanol [1].

Characterizations of Schiff Bases di Azomethine (A₁-A₃) (B₁) (B₂) (C₁) (C₂):

[A₁] 1,1'-(1,4-phenylene)bis(N-(4-chlorophenyl)methanimine): color yellow, state solid, yield 83%, M.p. 182-183°C, R_f = 0.62 (benzene: acetone) (9:1). FT-IR (ν cm⁻¹): 3057cm⁻¹ (C-H Aromatic), 1615cm⁻¹ (C=N), 1581-1477cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, CDCl₃): 8.48 (s, CH=N), 8.00-7.18 (m, H Aromatic). ¹³CNMR (400MHz, CDCl₃): 159.66 (CH=N), 150.11-120.31 (C Aromatic).

[A₂] 1,1'-(1,4-phenylene)bis(N-(4-bromophenyl)methanimine): color yellow, state solid, yield 88%, M.p. 121-122°C, R_f = 0.6 (benzene: acetone) (9:1). FT-IR (ν cm⁻¹): 3054cm⁻¹ (C-H Aromatic), 1611cm⁻¹ (C=N), 1562-1474cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 8.73 (s, CH=N), 8.09-7.28 (m, H Aromatic). ¹³CNMR (400MHz, DMSO-*d*₆): 161.39 (CH=N), 149.44-123.85 (C Aromatic).

[A₃] 1,1'-(1,4-phenylene)bis(N-(3-bromophenyl)methanimine): color greenish yellow, state solid, yield 91%, M.p. 178-179°C, R_f = 0.7 (benzene: acetone) (9:1). FT-IR (ν cm⁻¹): 3048cm⁻¹ (C-H Aromatic), 1620cm⁻¹ (C=N), 1560-1467cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 8.75 (s, CH=N), 8.16-7.32 (m, H Aromatic). ¹³CNMR (400MHz, DMSO-*d*₆): 162.20 (CH=N), 153.33-121.44 (C Aromatic).

[B₁] 1,1'-(1,3-phenylene)bis(N-(4-chlorophenyl)methanimine): color yellow, state solid, yield 83%, M.p. 108-109°C, R_f = 0.63 (benzene: acetone) (9:1). FT-IR (ν cm⁻¹): 3055cm⁻¹ (C-H Aromatic), 1621cm⁻¹ (C=N), 1577-1480cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz,

DMSO-*d*₆): 8.74 (s, $\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 8.55-7.33 (m, $\underline{\text{H}}$ Aromatic). ¹³CNMR (400MHz, DMSO-*d*₆): 161.38 ($\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 150.39-123.41 ($\underline{\text{C}}$ Aromatic).

[B₂] 1,1'-(1,3-phenylene)bis(N-(4-bromophenyl)methanimine): color greenish white, state solid, yield 69%, M.p. 123-124°C, R_f = 0.6 (benzene: acetone) (9:1). FT-IR (ν cm⁻¹): 3078cm⁻¹ (C-H Aromatic), 1626cm⁻¹ (C=N), 1577-1479cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 8.74 (s, $\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 8.52-7.27 (m, $\underline{\text{H}}$ Aromatic). ¹³CNMR (400MHz, DMSO-*d*₆): 161.43 ($\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 150.80-119.30 ($\underline{\text{C}}$ Aromatic).

[C₁] 1,1'-(1,2-phenylene)bis(N-(4-chlorophenyl)methanimine) [16]: color yellow, state solid, yield 88%, M.p. 171-172°C, R_f = 0.73 (benzene: acetone) (9:1). FT-IR (ν cm⁻¹): 3078cm⁻¹ (C-H Aromatic), 1611cm⁻¹ (C=N), 1587-1483cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 8.08 (s, $\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 7.63-7.36 (m, $\underline{\text{H}}$ Aromatic). ¹³CNMR (400MHz, DMSO-*d*₆): 153.47 ($\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 141.40-121.94 ($\underline{\text{C}}$ Aromatic).

[C₂] 1,1'-(1,2-phenylene)bis(N-(4-bromophenyl)methanimine): color orange, state solid, yield 82%, M.p. 185-186°C, R_f = 0.7 (benzene: acetone) (9:1). FT-IR (ν cm⁻¹): 3075cm⁻¹ (C-H Aromatic), 1611cm⁻¹ (C=N), 1576-1489cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 8.01 (s, $\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 7.60-7.34 (m, $\underline{\text{H}}$ Aromatic). ¹³CNMR (400MHz, DMSO-*d*₆): 149.99 ($\underline{\text{C}}\underline{\text{H}}=\underline{\text{N}}$), 140.84-122.32 ($\underline{\text{C}}$ Aromatic).

General Method for Preparing 1,3-Oxazepindion derivatives:

In a 50-mL reaction flask containing 0.001 moles (0.5g) of Schiff Bases di Azomethine in 15mL of hot-dried benzene. 0.002 moles of dissolved cyclic anhydrides (maleic and succinic anhydrides) in 15 mL of hot dried benzene were added. For 5–6 hours, the reaction mixture was left to reflux and then stirred at room temperature for an additional hour. The precipitate was filtered, and it was then recrystallized using dried benzene [1].

Characterizations of 1,3-Oxazepindione derivatives (DA₁-DA₃) (DB₁) (DB₂) (DC₁) (DC₂):

[DA₁] 2,2'-(1,4-phenylene)bis(3-(4-chlorophenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione): color pale yellow, state solid, yield 68%, M.p. 202-203°C, R_f = 0.58 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3076cm⁻¹ (C-H Aromatic), 3011cm⁻¹ (C-H alkene), 1701cm⁻¹ (C=O lactone), 1628cm⁻¹ (C=O lactam), 1522-1483cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 10.48 (s, O- $\underline{\text{C}}\underline{\text{H}}-\underline{\text{N}}$), 7.66-7.37 (m, $\underline{\text{H}}$ aromatic), 6.48-6.30 (m, $\underline{\text{C}}\underline{\text{H}}=\underline{\text{C}}\underline{\text{H}}$), ¹³CNMR (400MHz, DMSO-*d*₆): 167.37 ($\underline{\text{C}}\underline{\text{O}}-\underline{\text{O}}$), 163.80 ($\underline{\text{C}}\underline{\text{O}}-\underline{\text{N}}$), 138.08-129.20 ($\underline{\text{C}}$ aromatic), 130.65-127.79 ($\underline{\text{C}}\underline{\text{H}}=\underline{\text{C}}\underline{\text{H}}$), 121.43 (O- $\underline{\text{C}}\underline{\text{H}}-\underline{\text{N}}$).

[DA₂] 2,2'-(1,4-phenylene)bis(3-(4-bromophenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione): color pale yellow, state solid, yield 69%, M.p. 200-201°C, R_f = 0.55 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3082cm⁻¹ (C-H Aromatic), 3051cm⁻¹ (C-H alkene), 1702cm⁻¹ (C=O lactone), 1626cm⁻¹ (C=O lactam), 1504-1483cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 10.47 (s, O- $\underline{\text{C}}\underline{\text{H}}-\underline{\text{N}}$), 7.61-7.49 (m, $\underline{\text{H}}$ aromatic), 6.48-6.30 (m, $\underline{\text{C}}\underline{\text{H}}=\underline{\text{C}}\underline{\text{H}}$), ¹³CNMR (400MHz, DMSO-*d*₆): 167.37 ($\underline{\text{C}}\underline{\text{O}}-\underline{\text{O}}$), 163.82 ($\underline{\text{C}}\underline{\text{O}}-\underline{\text{N}}$), 138.51-129.20 ($\underline{\text{C}}$ aromatic), 130.65-122.05 ($\underline{\text{C}}\underline{\text{H}}=\underline{\text{C}}\underline{\text{H}}$), 115.84 (O- $\underline{\text{C}}\underline{\text{H}}-\underline{\text{N}}$).

[DA₃] 2,2'-(1,4-phenylene)bis(3-(3-bromophenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione): color pale green, state solid, yield 63%, M.p. 199-200°C, R_f = 0.66 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3079cm⁻¹ (C-H Aromatic), 3050cm⁻¹ (C-H alkene), 1712cm⁻¹ (C=O lactone), 1631cm⁻¹ (C=O lactam), 1530-1498cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 10.48 (s, O-CH-N), 7.98-7.26 (m, H aromatic), 6.48-6.31 (m, CH=CH), ¹³CNMR (400MHz, DMSO-*d*₆): 167.38 (CO-O), 164.01 (CO-N), 140.75-126.76 (C aromatic), 130.59-122.15 (CH=CH), 118.62 (O-CH-N).

[DB₁] 2,2'-(1,3-phenylene)bis(3-(4-chlorophenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione): color pale yellow, state solid, yield 64%, M.p. 206-207°C, R_f = 0.6 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3078cm⁻¹ (C-H Aromatic), 3055cm⁻¹ (C-H alkene), 1702cm⁻¹ (C=O lactone), 1629cm⁻¹ (C=O lactam), 1515-1485cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 10.48 (s, O-CH-N), 7.67-7.37 (m, H aromatic), 6.48-6.30 (m, CH=CH), ¹³CNMR (400MHz, DMSO-*d*₆): 167.36 (CO-O), 163.81 (CO-N), 138.08-129.19 (C aromatic), 130.64-127.80 (CH=CH), 121.44 (O-CH-N).

[DB₂] 2,2'-(1,3-phenylene)bis(3-(4-bromophenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione): color off-white, state solid, yield 65%, M.p. 209-210°C, R_f = 0.56 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3080cm⁻¹ (C-H Aromatic), 3052cm⁻¹ (C-H alkene), 1703cm⁻¹ (C=O lactone), 1626cm⁻¹ (C=O lactam), 1504-1484cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 10.47 (s, O-CH-N), 7.61-7.37 (m, H aromatic), 6.48-6.30 (m, CH=CH), ¹³CNMR (400MHz, DMSO-*d*₆): 167.37 (CO-O), 163.83 (CO-N), 138.50-129.33 (C aromatic), 130.65-122.05 (CH=CH), 115.85 (O-CH-N).

[DC₁] 2,2'-(1,2-phenylene)bis(3-(4-chlorophenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione): color pale yellow, state solid, yield 63%, M.p. 194-195°C, R_f = 0.62 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3078cm⁻¹ (C-H Aromatic), 3056cm⁻¹ (C-H alkene), 1704cm⁻¹ (C=O lactone), 1631cm⁻¹ (C=O lactam), 1523-1488cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 10.48 (s, O-CH-N), 7.67-7.07 (m, H aromatic), 6.49-6.30 (m, CH=CH), ¹³CNMR (400MHz, DMSO-*d*₆): 167.36 (CO-O), 163.81 (CO-N), 138.07-129.19 (C aromatic), 130.66-127.81 (CH=CH), 121.44 (O-CH-N).

[DC₂] 2,2'-(1,2-phenylene)bis(3-(4-bromophenyl)-2,3-dihydro-1,3-oxazepine-4,7-dione): color pale yellow, state solid, yield 66%, M.p. 205-206°C, R_f = 0.59 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3082cm⁻¹ (C-H Aromatic), 3052cm⁻¹ (C-H alkene), 1705cm⁻¹ (C=O lactone), 1628cm⁻¹ (C=O lactam), 1504-1485cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-*d*₆): 10.49 (s, O-CH-N), 7.62-7.23 (m, H aromatic), 6.49-6.31 (m, CH=CH), ¹³CNMR (400MHz, DMSO-*d*₆): 167.36 (CO-O), 163.83 (CO-N), 138.49-128.79 (C aromatic), 130.68-121.82 (CH=CH), 115.88 (O-CH-N).

Characterizations of 1,3-Oxazepindion derivatives (EA₁-EA₃) (EB₁) (EB₂) (EC₁) (EC₂):

[EA₁] 2,2'-(1,4-phenylene)bis(3-(4-chlorophenyl)-1,3-oxazepane-4,7-dione): color off-white, state solid, yield 68%, M.p. 172-173°C, R_f = 0.54 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3046cm⁻¹ (C-H Aromatic), 2965-2842cm⁻¹ (C-H aliphatic), 1693cm⁻¹ (C=O lactone), 1656cm⁻¹ (C=O lactam), 1593-1527cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-

d6): 10.11 (s, O-CH-N), 7.62-7.34 (m, H aromatic), 2.59-2.53 (m, CH₂-CH₂), ¹³CNMR (400MHz, DMSO-d₆): 174.27 (C=O), 170.73 (C=O-N), 138.71-123.46 (C aromatic), 31.49-29.17 (CH₂-CH₂), 120.86 (O-CH-N).

[EA₂] 2,2'-(1,4-phenylene)bis(3-(4-bromophenyl)-1,3-oxazepane-4,7-dione): color pale yellow, state solid, yield 71%, M.p. 198-199°C, R_f = 0.51 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3080cm⁻¹ (C-H Aromatic), 2980-2872cm⁻¹ (C-H aliphatic), 1696cm⁻¹ (C=O lactone), 1655cm⁻¹ (C=O lactam), 1563-1503cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-d₆): 10.10 (s, O-CH-N), 8.15-7.30 (m, H aromatic), 2.50-2.41 (m, CH₂-CH₂), ¹³CNMR (400MHz, DMSO-d₆): 174.11 (C=O), 170.26 (C=O-N), 138.55-123.85 (C aromatic), 31.60-29.34 (CH₂-CH₂), 119.43 (O-CH-N).

[EA₃] 2,2'-(1,4-phenylene)bis(3-(3-bromophenyl)-1,3-oxazepane-4,7-dione): color pale yellow, state solid, yield 66%, M.p. 164-165°C, R_f = 0.64 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3069cm⁻¹ (C-H Aromatic), 2959-2874cm⁻¹ (C-H aliphatic), 1695cm⁻¹ (C=O lactone), 1656cm⁻¹ (C=O lactam), 1583-1529cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-d₆): 10.10 (s, O-CH-N), 8.09-7.24 (m, H aromatic), 2.57-2.50 (m, CH₂-CH₂), ¹³CNMR (400MHz, DMSO-d₆): 174.23 (C=O), 170.99 (C=O-N), 138.83-123.84 (C aromatic), 31.51-29.11 (CH₂-CH₂), 118.06 (O-CH-N).

[EB₁] 2,2'-(1,3-phenylene)bis(3-(4-chlorophenyl)-1,3-oxazepane-4,7-dione): color white, state solid, yield 67%, M.p. 160-165°C, R_f = 0.55 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3045cm⁻¹ (C-H Aromatic), 2936-2840cm⁻¹ (C-H aliphatic), 1693cm⁻¹ (C=O lactone), 1656cm⁻¹ (C=O lactam), 1593-1527cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-d₆): 10.10 (s, O-CH-N), 7.63-7.32 (m, H aromatic), 2.58-2.52 (m, CH₂-CH₂), ¹³CNMR (400MHz, DMSO-d₆): 174.26 (C=O), 170.73 (C=O-N), 138.71-126.88 (C aromatic), 31.49-29.16 (CH₂-CH₂), 120.86 (O-CH-N).

[EB₂] 2,2'-(1,3-phenylene)bis(3-(4-bromophenyl)-1,3-oxazepane-4,7-dione): color white, state solid, yield 66%, M.p. 191-192°C, R_f = 0.54 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3068cm⁻¹ (C-H Aromatic), 2972-2870cm⁻¹ (C-H aliphatic), 1695cm⁻¹ (C=O lactone), 1661cm⁻¹ (C=O lactam), 1588-1518cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-d₆): 10.10 (s, O-CH-N), 7.64-7.37 (m, H aromatic), 2.58-2.53 (m, CH₂-CH₂), ¹³CNMR (400MHz, DMSO-d₆): 174.25 (C=O), 170.76 (C=O-N), 139.12-121.26 (C aromatic), 31.52-29.15 (CH₂-CH₂), 114.88 (O-CH-N).

[EC₁] 2,2'-(1,2-phenylene)bis(3-(4-chlorophenyl)-1,3-oxazepane-4,7-dione): color Pale yellow, state solid, yield 63%, M.p. 150-151°C, R_f = 0.58 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3027cm⁻¹ (C-H Aromatic), 2980-2868cm⁻¹ (C-H aliphatic), 1685cm⁻¹ (C=O lactone), 1655cm⁻¹ (C=O lactam), 1560-1506cm⁻¹ (C=C Aromatic), ¹HNMR (400MHz, DMSO-d₆): 10.11 (s, O-CH-N), 7.62-7.33 (m, H aromatic), 2.60-2.55 (m, CH₂-CH₂), ¹³CNMR (400MHz, DMSO-d₆): 174.09 (C=O), 170.74 (C=O-N), 138.12-122.03 (C aromatic), 31.45-29.20 (CH₂-CH₂), 120.85 (O-CH-N).

[EC₂] 2,2'-(1,2-phenylene)bis(3-(4-bromophenyl)-1,3-oxazepane-4,7-dione): color off-white, state solid, yield 64%, M.p. 162-163°C, R_f = 0.57 (n-hexane: ethyl acetate) (6:4). FT-IR (ν cm⁻¹): 3028cm⁻¹ (C-H Aromatic), 2978-2869cm⁻¹ (C-H aliphatic), 1689cm⁻¹ (C=O

lactone), 1656 cm^{-1} (C=O lactam), 1588-1524 cm^{-1} (C=C Aromatic), $^1\text{H-NMR}$ (400MHz, DMSO- d_6): 10.10 (s, O- CH-N), 7.57-7.37 (m, H aromatic), 2.58-2.53 (m, $\text{CH}_2\text{-CH}_2$), $^{13}\text{C-NMR}$ (400MHz, DMSO- d_6): 174.25 (C=O-O), 170.76 (C=O-N), 139.12-121.28 (C aromatic), 31.51-29.15 ($\text{CH}_2\text{-CH}_2$), 114.87 (O- CH-N).

RESULTS AND DISCUSSION

Schiff Bases di Azomethine ($\text{A}_1\text{-A}_3$) (B_1) (B_2) (C_1) (C_2) were prepared by thermal condensation between Terephthalaldehyde, Isophthalaldehyde, or *o*-Phthalaldehyde and aromatic amines (halo aniline) with 5 drops of glacial acetic acid and use absolute ethanol as a solvent. Scheme 1 shows the paths for preparing Schiff Bases di Azomethine, and Scheme 2 shows the mechanism of reaction for the preparation of Schiff Bases di Azomethine [10].

All FT-IR spectra for synthesized Schiff Bases di Azomethine showed the disappearance of the absorption bands of the (C=O) group in aldehydes and the (-NH₂) group in amines and the appearance of the absorption bands of the (C-H) aromatic at (3078–3048 cm^{-1}), Azomethine groups (C=N) at (1617–1612 cm^{-1}), and (C=C) aromatic at (1587–1467 cm^{-1}) [17] [18]. Figure 1 shows FT-IR spectra for [A_2] compound.

All $^1\text{H-NMR}$ spectra of the synthesized Schiff Bases di Azomethine showed signals of (N=C-H) (2H, s) at ($\delta=8.75\text{-}8.01\text{ppm}$) and aromatic protons (12H, m) at ($\delta=8.55\text{-}7.18\text{ppm}$) [19]. Figure 2 shows $^1\text{H-NMR}$ spectra for [A_1] compound.

All $^{13}\text{C-NMR}$ spectra of the synthesized Schiff Bases di Azomethine showed signals of (N=C-H) (2C) at ($\delta=162.20\text{-}149.99\text{ppm}$) and aromatic carbons (18C) at ($\delta=153.33\text{-}119.30\text{ppm}$). Figure 3 shows $^{13}\text{C-NMR}$ spectra for [B_1] compound.

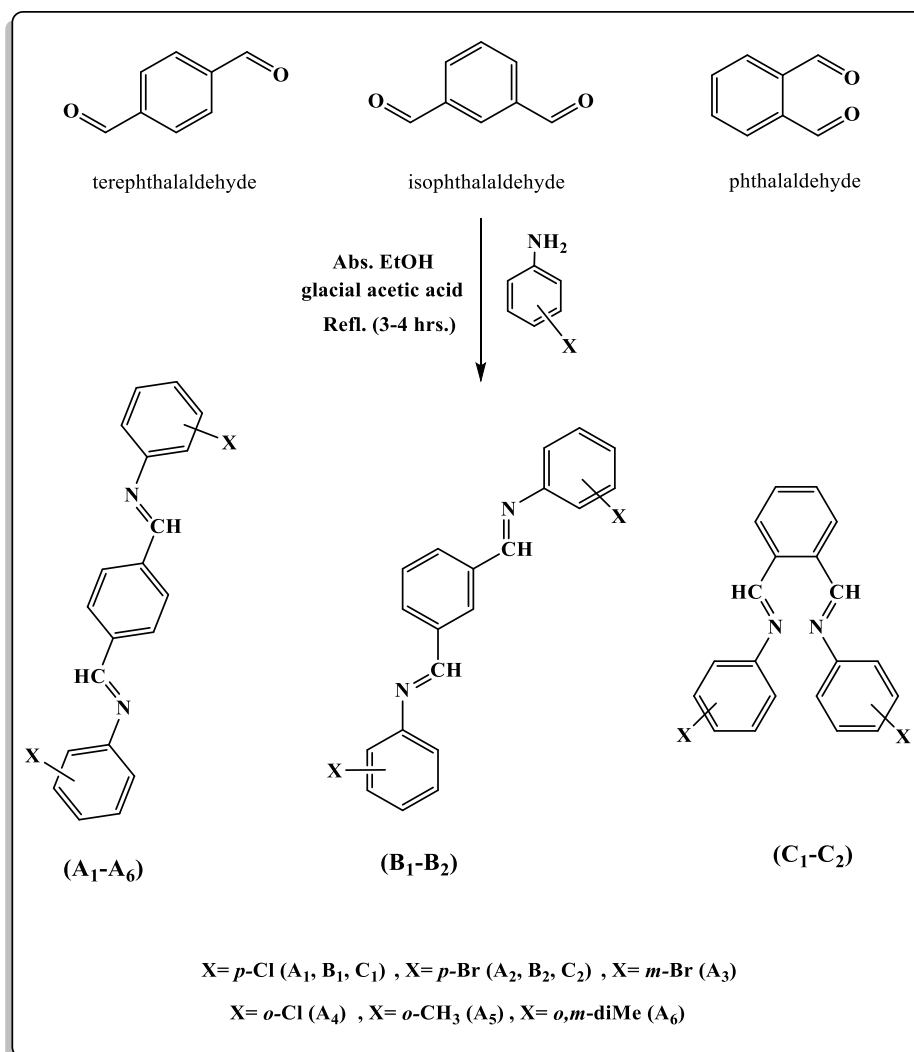
1,3-Oxazepindione derivatives were obtained by cycloaddition reactions between prepared Schiff Bases di Azomethine with Maleic anhydride or Succinic anhydride using dried benzene as a solvent. Schemes 3a and 3b show the paths for preparing 1,3-Oxazepindione derivatives, and Scheme 4 shows the mechanism of reaction for the preparation of 1,3-Oxazepindione derivatives [20].

All FT-IR spectra for synthesized 1,3-Oxazepindione derivatives showed the disappearance of the absorption band of Azomethine groups (C=N) at (1617–1612 cm^{-1}) and the appearance of the absorption bands of the (C-H) aromatic at (3082–3027 cm^{-1}), (CH=CH) at (3056–3011 cm^{-1}), (CH₂-CH₂) at (2980–2840 cm^{-1}), (C=O) lactone at (1712–1685 cm^{-1}), (C=O) lactam at (1655–1626 cm^{-1}) and (C=C) aromatic at (1593–1483 cm^{-1}) (17) (18). Figure 4 shows FT-IR spectra for [DB_2] compound.

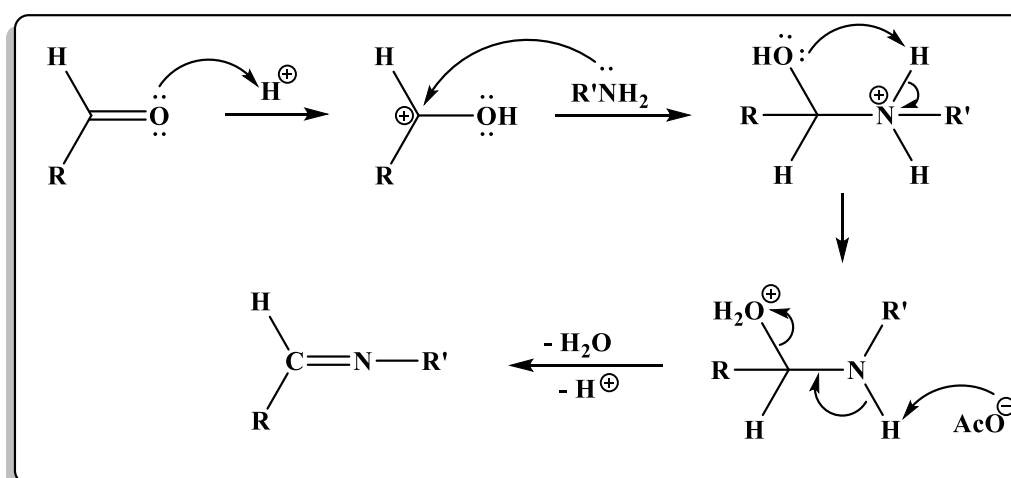
All $^1\text{H-NMR}$ spectra of synthesized 1,3-Oxazepindione derivatives showed signals of (N- CH-O) (2H, s) at ($\delta=10.49\text{-}10.10\text{ppm}$), aromatic protons (12H, m) at ($\delta=8.15\text{-}7.07\text{ppm}$), (CH=CH) (4H, d) at ($\delta=6.49\text{-}6.30\text{ppm}$) and (CH₂-CH₂) (8H, t) at ($\delta=2.60\text{-}2.41\text{ppm}$) (19). Figure 5 shows $^1\text{H-NMR}$ spectra for [EC_2] compound.

All $^{13}\text{C-NMR}$ spectra of the synthesized 1,3-Oxazepindione derivatives showed signals of (O=C-O) (2C) at ($\delta=174.27\text{-}167.36\text{ppm}$), (O=C-N) (2C) at ($\delta=170.99\text{-}163.80\text{ppm}$), aromatic carbons (18C) at ($\delta=140.75\text{-}121.26\text{ppm}$), (N- CH-O) (2C) at ($\delta=121.44\text{-}114.87\text{ppm}$),

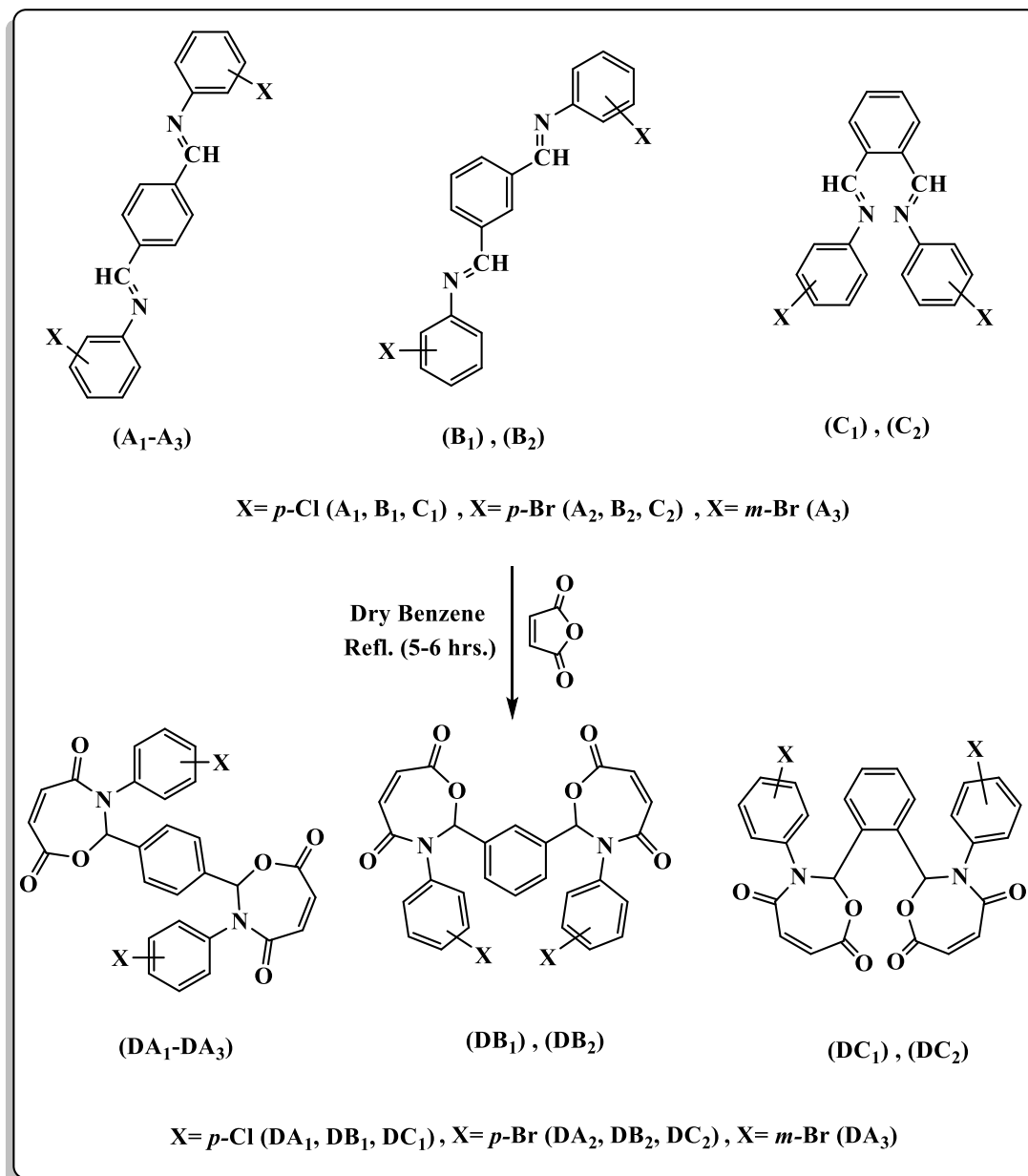
($\underline{\text{C}}\text{H}_2=\underline{\text{C}}\text{H}_2$) (4C) at ($\delta=130.68-121.82\text{ppm}$), and ($\underline{\text{C}}\text{H}_2-\underline{\text{C}}\text{H}_2$) (4C) at ($\delta=31.60-29.11\text{ppm}$). Figure 6 shows ^{13}C -NMR spectra for [DA₁] compound.



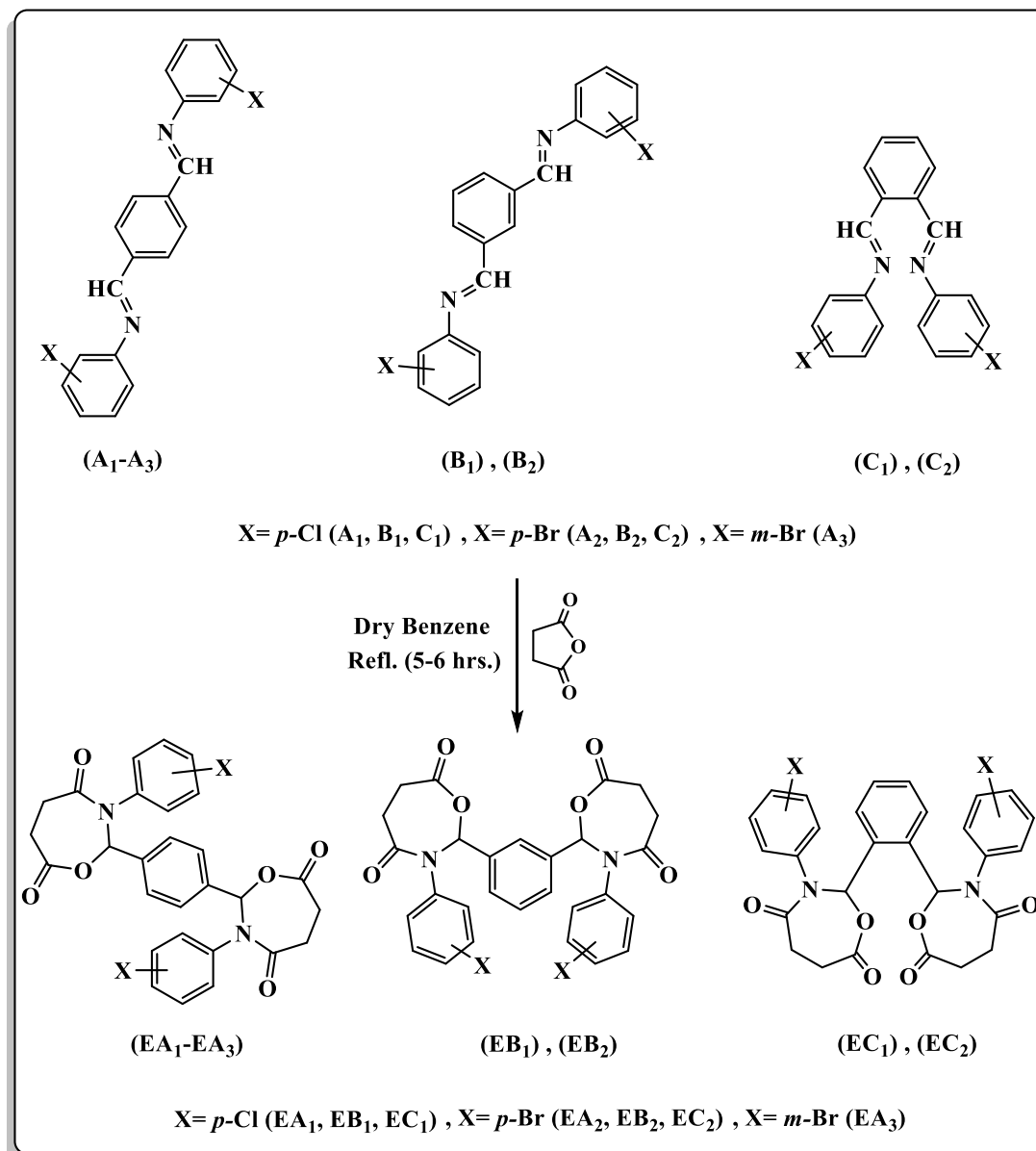
Scheme 1: the paths for preparing Schiff Bases di Azomethine [A][B][C].



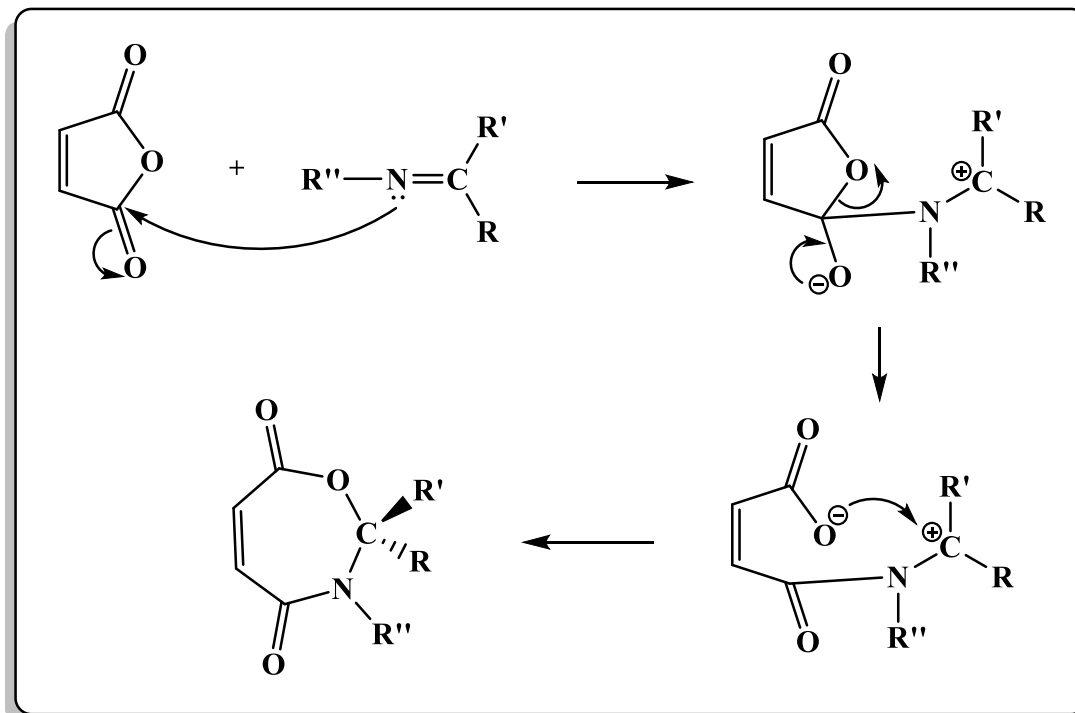
Scheme 2: the mechanism of reaction for the preparation of Schiff Bases di Azomethine.



Schemes 3a: the paths for preparing 1,3-Oxazepindione derivatives [D].



Schemes 3b: the paths for preparing 1,3-Oxazepindione derivatives [E]



Scheme 4: the mechanism of reaction for the preparation of 1,3-Oxazepindione derivatives.

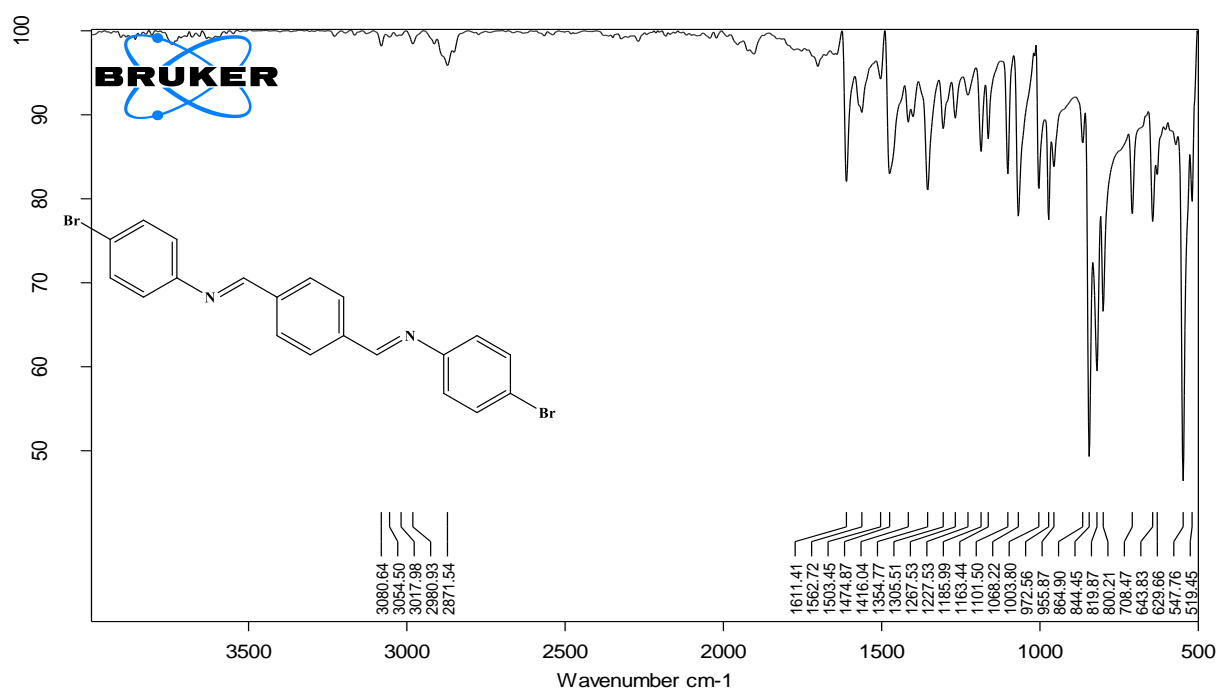


Fig. 1: shows FT-IR spectra for [A₂].

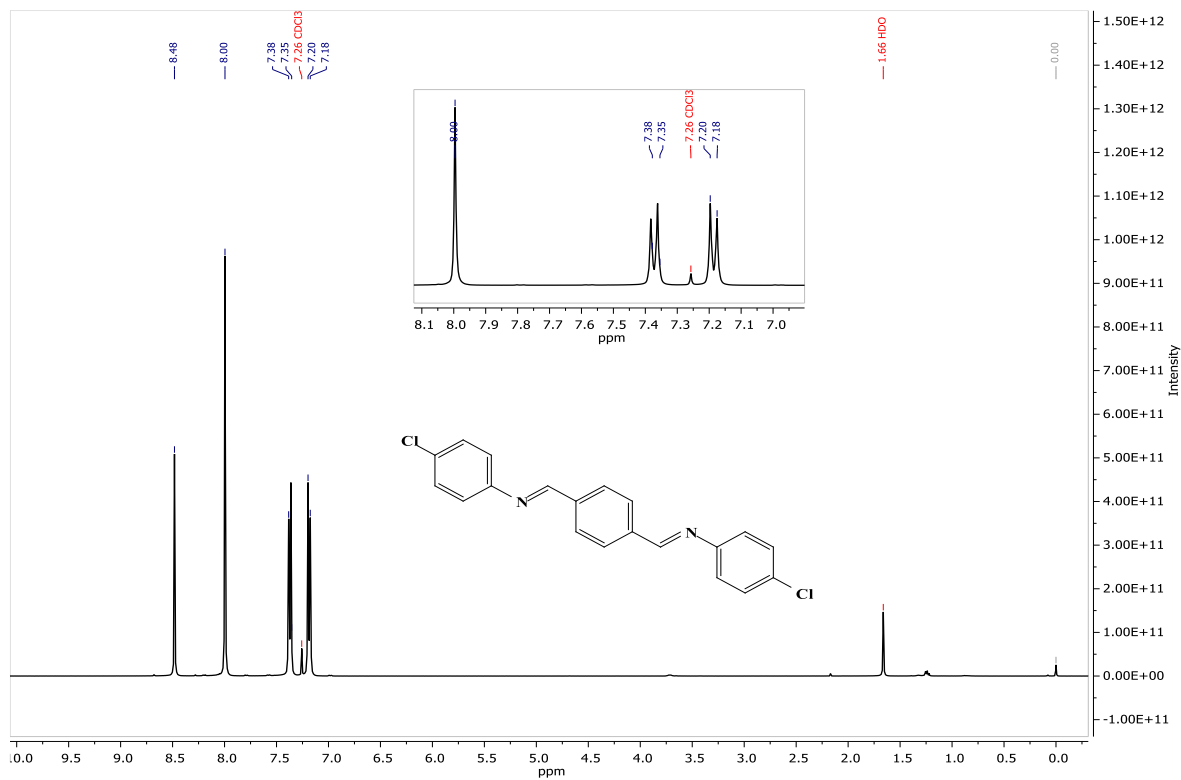


Fig. 2: shows $^1\text{H-NMR}$ spectra for [A₁].

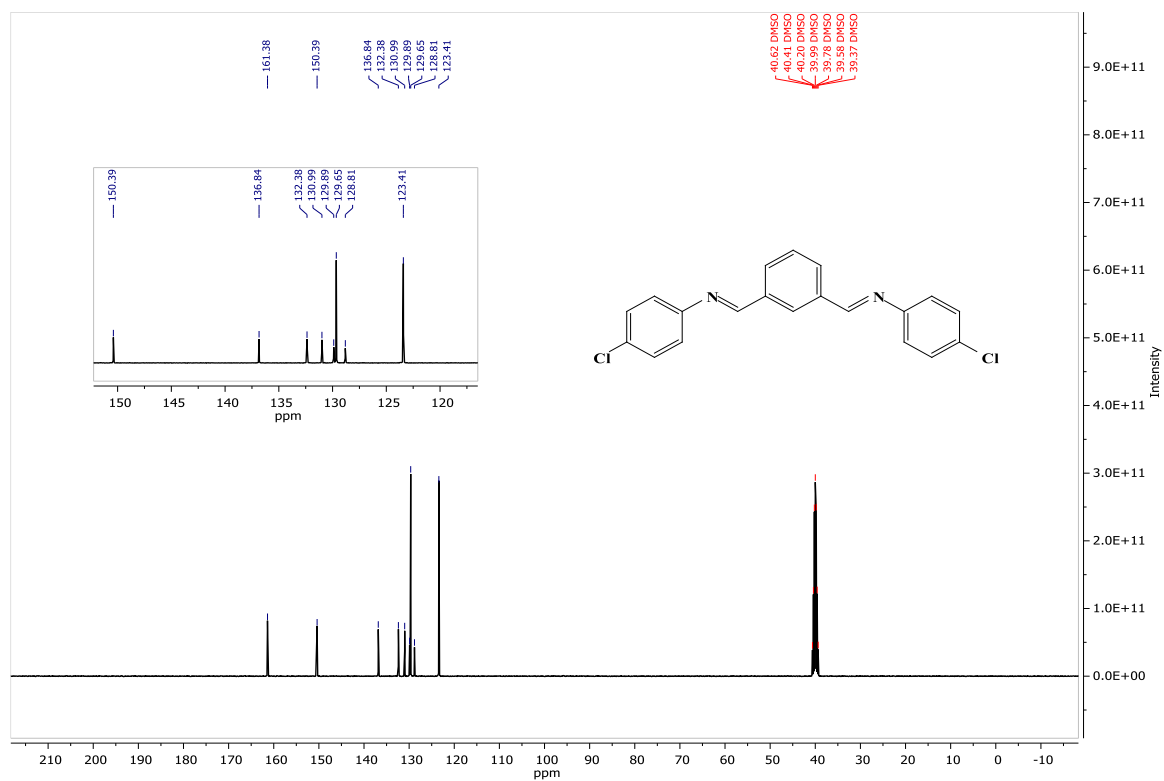


Fig. 3: shows $^{13}\text{C-NMR}$ spectra for [B₁].

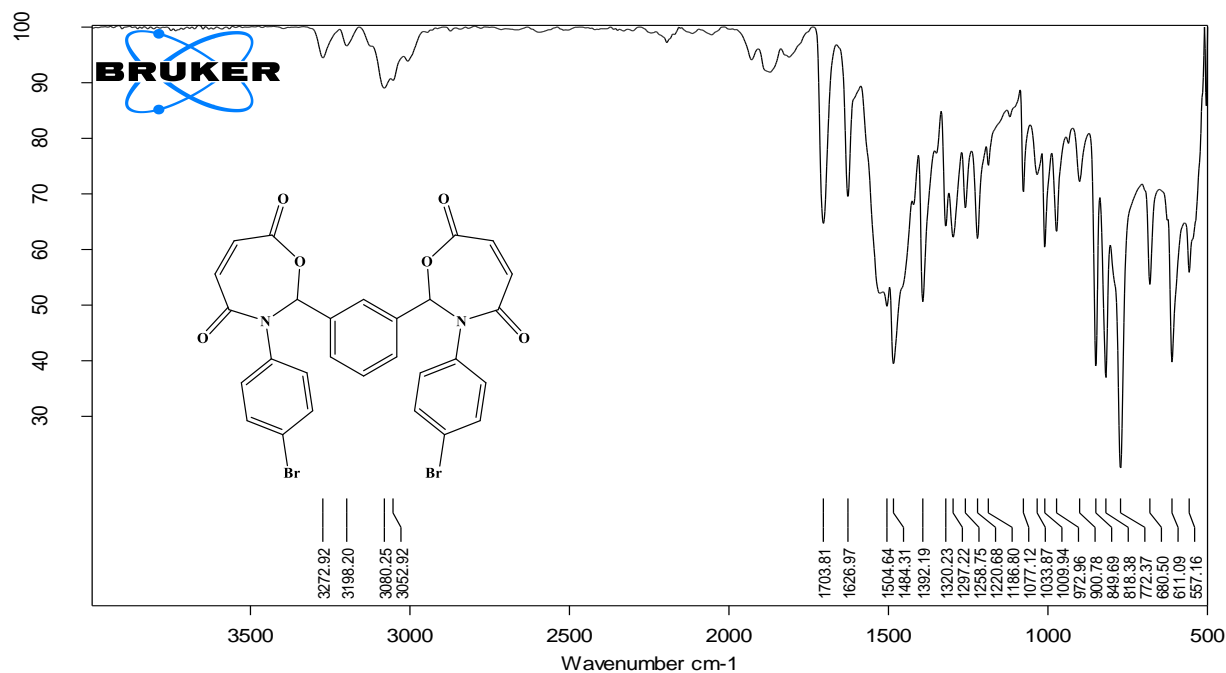


Fig. 4: shows FT-IR spectra for [DB₂].

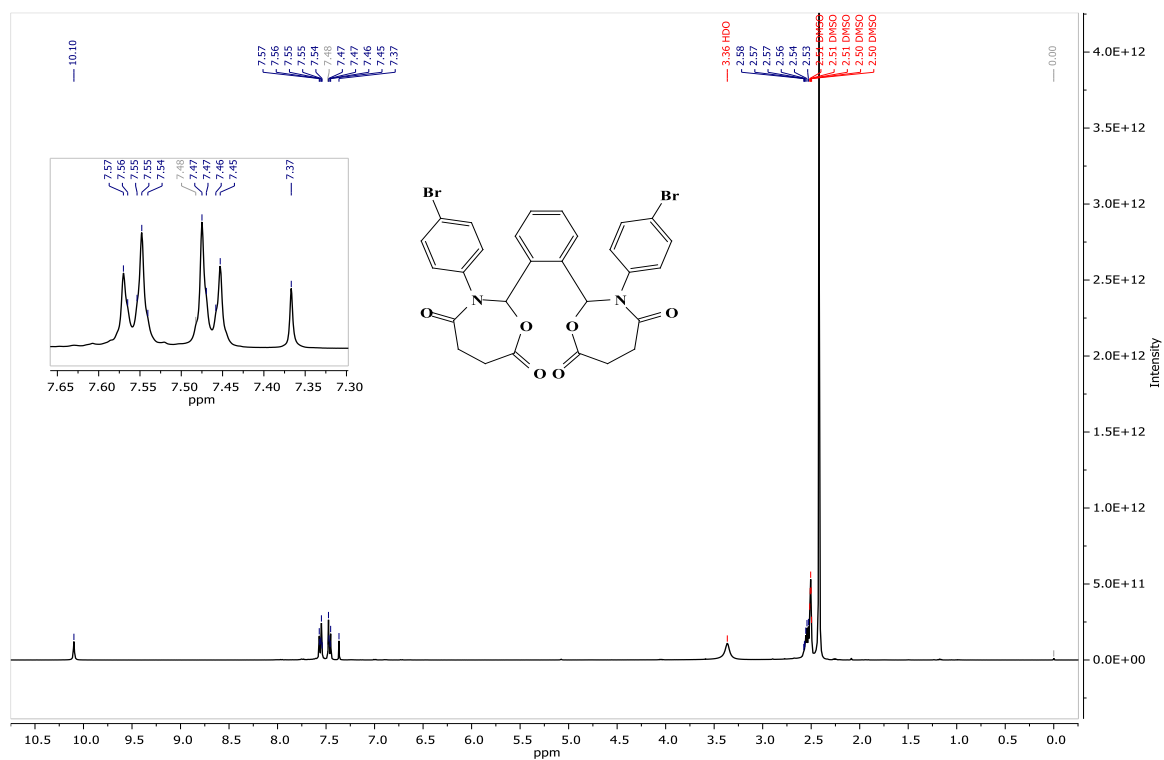


Fig. 5: shows ¹H-NMR spectra for [EC₂].

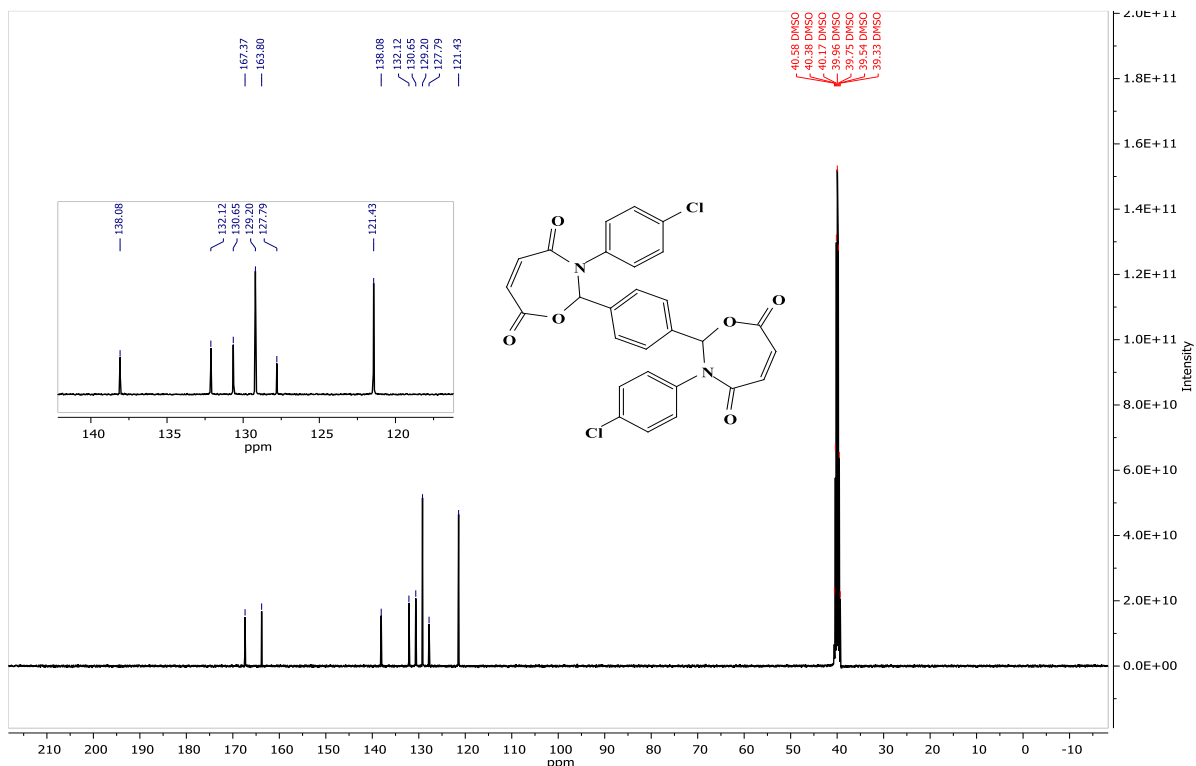


Fig. 6: shows ^{13}C -NMR spectra for [DA₁].

CONCLUSION

All compounds required for this scientific study were successfully prepared. Nuclear magnetic resonance spectroscopy (^1H NMR and ^{13}C NMR) and infrared (FT-IR) spectroscopy were used to characterize the prepared compounds. Using thin layer chromatography (TLC), all reactions were monitored.

References

1. Mahdi, M. S., & Kshash, A. H. (2020). Synthesis and characterization of two moieties of-1, 3-oxazepine-1, 5-dione compounds via 4, 4'-diaminodiphenylmethane imines as precursors. *Biochemical & Cellular Archives*, 20(2). <https://connectjournals.com/03896.2020.20.4719>.
2. Kumar, K. S., & Aravindakshan, K. K. (2021). Synthesis, cytotoxic, anticancer and antimicrobial activities of some metal complexes of a novel tetradentate Schiff base ligand, (E)-3-((2-((E)-(1-(2-hydroxyphenyl) ethylidene) amino) ethyl) imino)-N-phenylbutanamide. *Results in Chemistry*, 3, 100129. doi.org/10.1016/j.rechem.2021.100129.
3. Al-Labban, H. M. Y., Sadiq, H. M., & Aljanaby, A. A. J. (2019, September). Synthesis, Characterization and study biological activity of some Schiff bases derivatives from 4-amino antipyrine as a starting material. *In Journal of Physics: Conference Series* (Vol. 1294, No. 5, p. 052007). IOP Publishing. [doi:10.1088/1742-6596/1294/5/052007](https://doi.org/10.1088/1742-6596/1294/5/052007).
4. Omidi, S., & Kakanejadifard, A. (2020). A review on biological activities of Schiff base, hydrazone, and oxime derivatives of curcumin. *RSC advances*, 10(50), 30186-30202. <https://doi.org/10.1039/D0RA05720G>.

5. Mermer, A., Keles, T., & Sirin, Y. (2021). Recent studies of nitrogen containing heterocyclic compounds as novel antiviral agents: *A review. Bioorganic Chemistry*, 114, 105076. doi.org/10.1016/j.bioorg.2021.105076.
6. Jassim, [W. K.](#) (2018). Synthesis and antibacterial study for some heterocyclic compounds. *Kerbala journal of pharmaceutical sciences*, (14), 82-92. <https://www.iasj.net/iasj/article/146123>.
7. Al-Mulla, A. (2017). A review: biological importance of heterocyclic compounds. *Der Pharma Chemica*, 9(13), 141-147. <https://www.derpharmachemica.com/abstract/a-review-biological-importance-of-heterocyclic-compounds-12742.html>.
8. Omar, F. A., Hamad, A. S., & Taha, N. I. (2022). Synthesis, Characterization and Evaluation Antibacterial Activity of Some (1, 3-Oxazepine-4, 7-dione and 1, 3-Benzooxazepine-4, 7-dione) Derived from Sulphamethoxazole using Irradiation Method. *Kirkuk University Journal-Scientific Studies*, 17(2), 27-35. <https://doi.org/10.32894/kujss.2022.131552.1048>.
9. Nadr, R. B., & Abdulrahman, B. S. (2023). Synthesis and Characterization of a New Series of [1, 3]-Oxazepine Compounds from Heterocyclic Schiff Bases. *Zanco Journal of Pure and Applied Sciences*, 35(2), 197-210. <http://dx.doi.org/10.21271/ZJPAS.35.2.21>.
10. Shukkur, A. (2020). Synthesis of Six and Seven-membered Heterocyclic Molecules Containing an Adamantyl Fragment and an X-ray Crystal Structure of (E)-N-(adamantan-1-yl)-1-(3-nitrophenyl) methanimine. *Baghdad Science Journal*, 17(1 (Suppl.)), 0272-0272. doi.org/10.21123/bsj.2020.17.1(Suppl.).0272.
11. Allamy, A. K. N., & Mejbil, S. A. (2022). Preparation, characterization and biological activity of some new seven-membered heterocyclic compounds. *World Journal of Advanced Research and Reviews*, 15(1), 662-678. doi.org/10.30574/wjarr.2022.15.1.0744.
12. Muslim, R. F., Tawfeeq, H. M., Owaid, M. N., & Abid, O. H. (2018). Synthesis, characterization and evaluation of antifungal activity of seven-membered heterocycles. *Acta Pharmaceutica Scientia*, 56(2). DOI: 10.23893/1307-2080.APS.05610.
13. Mousa, E. F., & Jassim, I. K. (2021). Synthesis, characterization, and study the biological activity of some schiff's bases, and 1, 3-oxazepine compounds derived from sulfamethoxazole drug. *Iraqi Journal of Market Research and Consumer Protection*, 13(1), 43-54. [http://dx.doi.org/10.28936/jmracpc13.1.2021.\(5\)](http://dx.doi.org/10.28936/jmracpc13.1.2021.(5)).
14. Khelifi, I., Pecnard, S., Bernadat, G., Bignon, J., Levaique, H., Dubois, J., ... & Alami, M. (2020). Synthesis and Anticancer Properties of Oxazepines Related to Azaisoerianin and IsoCoQuines. *ChemMedChem*, 15(16), 1571-1578. <https://doi.org/10.1002/cmdc.202000197>.
15. Thummala, Y., Raju, C. E., Purnachandar, D., Sreenivasulu, G., Doddi, V. R., & Karunakar, G. V. (2020). Gold-Catalyzed Regioselective Synthesis of Pyrazolo [1, 4] oxazepines via Intramolecular 7-endo-dig Cyclization. *European Journal of Organic Chemistry*, 2020(24), 3560-3567. <https://doi.org/10.1002/ejoc.201901852>.
16. MAHDI, S. H., & KAREM, L. K. A. (2020). Synthesis, Physicochemical Studies and biological estimation of new mixed ligand complexes from hetrocyclic compounds. *International Journal of Pharmaceutical Research*, (1). <https://doi.org/10.31838/ijpr/2020.SP1.267>.
17. Silverstein, R. M., Webster, F. X., Kiemle, D. J., Kiemle, D. J., & Webster, F. X. (2005). Spectrometric identification of organic compounds (7th ed.). John Wiley & Sons.
18. Pavia, D. L., Lampman, G. M., Kriz, G. S., & Vyvyan, J. R. (2015). Introduction to spectroscopy (Fifth edition. ed.). Cengage Learning.

19. Claridge, T. D. W. (2016). High-Resolution NMR Techniques in Organic Chemistry. Elsevier Science.
20. Ahmed, A., Mahdi, S., Hussein, A., Hamed, A., & Yousif, E. (2015). Antibacterial study of some oxazepine derivatives. *Al-Nahrain Journal of Science*, 18(4), 22-26. <https://www.iasj.net/iasj/article/107267>.

تحضير وتشخيص مشتقات 1،3-اوكسازيبندايون عن طريق إيمينات ثنائي فورميل البنزين كمادة أولية

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الخلاصة:

في هذا البحث، حضرت مشتقات 1،3-اوكسازيبندايون [D] و [E] عن طريق تفاعل قواعد شف ثنائية الازوميثين [[B، A]] و [C] مع أنهريد المالك وأنهدريد السكسنيك بوجود البنزين الجاف كمذيب. و حضرت قواعد شف ثنائية الازوميثين [A]، [B] و [C] عن طريق تفاعل ترفثالديهايد او ايزوفثالديهايد او اورثوفثالديهايد مع 5 قطرات من حاض الخليك الثلجي كعامل مساعد ومشتقات الانيلين الهالوجينية في الايثانول المطلق كمذيب. تم تشخيص المركبات المحضرة واثباتها باستخدام مطيافية الاشعة تحت الحمراء والرنين النووي المغناطيسي. جميع تفاعلات المركبات المحضرة تمت متابعتها عن طريق تنقية كروماتوغرافيا الطبقة الرقيقة.

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الكلمات المفتاحية:

1،3-اوكسازيبين، قواعد شف، إيمينات، أنهريد، الأنيلين.

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